## Chemistry

## Lecture 3

Q.1 15 mL of water and 10 mL of sulphuric acid are to be mixed in a beaker

(i) State the method that should be followed with reason.

(ii)What is this process called?

- Q.2 Classify the following salts into acidic, basic and neutral: Potassium sulphate, ammonium chloride, sodium carbonate, sodium chloride
- Q.3 (a)What is an alkali? Give an example.

(b) Why do HCl, *HNO*<sub>3</sub>, etc. show acidic characters in aqueous solutions while solutions of compounds like alcohol and glucose do not show acidic character?

Q.4 What would be the colour of red litmus in a solution of sodium carbonate?

